

### Surface Energy Data for PCP: Polychloroprene, CAS # 9010-98-4

Source <sup>(a)</sup>	Mst. Type <sup>(b)</sup>	Data <sup>(c)</sup>	Comments <sup>(d)</sup>
Crocker, 1969 <sup>(111)</sup>	Critical ST	$\gamma_c = 38 \text{ mJ/m}^2$ ; no temp cited	Test liquids not known.
Shafrin, 1975 <sup>(297)</sup>	Critical ST	$\gamma_c = 38 \text{ mJ/m}^2$ ; 20°C	Test liquids not known.
Wu, 1971 <sup>(29)</sup>	From polymer melt	$\gamma_s = 43.6 \text{ mJ/m}^2$ ( $\gamma_s^d = 38.7$ , $\gamma_s^p = 4.9$ ); 20°C	Measurement by pendant drop of polymer melt extrapolated to 20°C; polarity calculated from interfacial tension with PE by geometric mean equation. $M_v = 30,000$ .
Wu, 1971 <sup>(29)</sup>	From polymer melt	$\gamma_s = 43.6 \text{ mJ/m}^2$ ( $\gamma_s^d = 38.9$ , $\gamma_s^p = 4.7$ ); 20°C	Measurement by pendant drop of polymer melt extrapolated to 20°C; polarity calculated from interfacial tension with PE by harmonic mean equation. $M_v = 30,000$ .
Sewell, 1971 <sup>(193)</sup>	Calculated	$\gamma_s = 43.3 \text{ mJ/m}^2$ ; no temp cited	Calculated from parachor and cohesive energy.
Sewell, 1971 <sup>(193)</sup>	Calculated	$\gamma_s = 35.3 \text{ mJ/m}^2$ ; no temp cited	Calculated by least squares from cohesive energy and molar volume.
Wu, 1974 <sup>(47)</sup>	Calculated	$\gamma_s = 44.2 \text{ mJ/m}^2$ ; 20°C	Calculated from parachor and molecular weight.
Pritykin, 1986 <sup>(199)</sup>	Calculated	$\gamma_s = 38.4 \text{ mJ/m}^2$ ; no temp cited	Calculated from cohesion parameters and monomer refractometric characteristics, equation 1.
Pritykin, 1986 <sup>(199)</sup>	Calculated	$\gamma_s = 37.7 \text{ mJ/m}^2$ ; no temp cited	Calculated from cohesion parameters and monomer refractometric characteristics, equation 2.